

Photoredox Catalysis

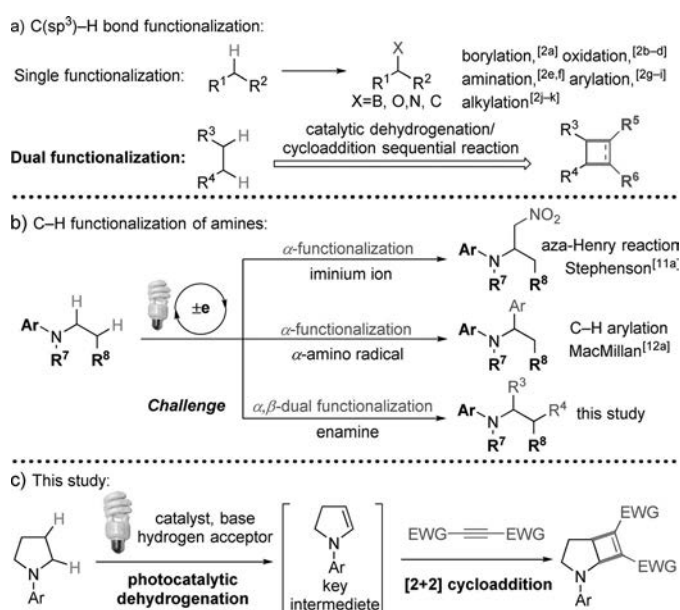
International Edition: DOI: 10.1002/anie.201710523
German Edition: DOI: 10.1002/ange.201710523Dual C(sp³)-H Bond Functionalization of N-Heterocycles through Sequential Visible-Light Photocatalyzed Dehydrogenation/[2+2] Cycloaddition Reactions

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Abstract: Herein we describe a mild method for the dual C(sp³)-H bond functionalization of saturated nitrogen-containing heterocycles through a sequential visible-light photocatalyzed dehydrogenation/[2+2] cycloaddition procedure. As a complementary approach to the well-established use of iminium ion and α -amino radical intermediates, the elusive cyclic enamine intermediates were effectively generated by photoredox catalysis under mild conditions and efficiently captured by acetylene esters to form a wide array of bicyclic amino acid derivatives, thus enabling the simultaneous functionalization of two vicinal C(sp³)-H bonds.

The direct functionalization of C-H bonds is one of the most challenging yet highly desirable goals in modern organic synthesis,^[1] and developments^[2] in direct transformations of ubiquitous C(sp³)-H bonds, including borylation, oxidation, amination, arylation, and alkylation, have attracted much attention. However, the vast majority of examples have focused on the functionalization of a single C(sp³)-H bond, and dual C(sp³)-H bond functionalization in a single step is still unknown. Encouraged by our previous study, in which we realized the dual functionalization of both a C(sp³)-H and a C(sp²)-H bond through visible-light induced oxidation and an aza-Diels-Alder reaction,^[3] we planned to explore the feasibility of dual C(sp³)-H bond functionalization. Catalytic dehydrogenation^[4] has provided a new method for activating two C(sp³)-H bonds, since the reactive alkene intermediate generated by this process could be utilized in a secondary reaction to forge new carbon frameworks.^[5] [2+2] cycloaddition is one of the most powerful and widely applied transformations in organic chemistry^[6] and plays a remarkable role in building strained and unusual molecular architectures that cannot be accessed through other pathways.^[7] Recently, Zhou and co-workers developed an unprecedented catalytic

enantioselective Heck annulation of propargylic acetates and cycloalkenes to obtain highly strained, fused cyclobutenes.^[8] On the basis of all these factors, we decided to tackle the aforementioned difficulty by merging catalytic dehydrogenation with [2+2] cycloaddition in a sequential process (Scheme 1 a).



Scheme 1. Dual C(sp³)-H functionalization strategy. EWG = electron-withdrawing group.

In recent years, visible-light photoredox catalysis,^[9] another powerful technology in organic synthesis, has had a profound impact on the C-H functionalization of saturated N-heterocycles, which represent a privileged motif in pharmaceuticals and natural products.^[10] Although a variety of synthetically useful iminium ion^[11] and α -amino radical^[12] intermediates have been effectively generated by photoredox catalysis and successfully applied to various α -functionalization reactions of amines,^[13] enamines as promising intermediates for α,β -functionalization of amines are rarely reported (Scheme 1 b). Besides the general difficulties associated with C(sp³)-H functionalization (e.g., high bond energy and similar reactivities), it faces greater challenges in the context of the dual C-H functionalization of saturated N-heterocycles: 1) the need to generate sufficient amounts of the reactive cyclic enamine intermediate and avoid the formation

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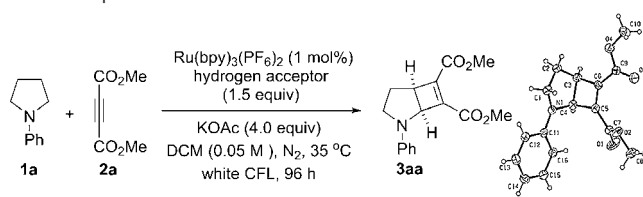
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of more stable aromatic heterocycles,^[14] and 2) the strong competition of the well-established α -functionalization^[13] of iminium ions^[11] and α -amino radicals^[12] with this process. Herein, we describe a mild method to generate the highly reactive cyclic enamine intermediate by a visible-light photocatalytic dehydrogenation process, and then dual C(sp³)-H bond functionalization through coupling with traditional [2+2] cycloaddition, to furnish a variety of highly strained bicyclic amino acid derivatives, which are rather difficult to obtain by other methods (Scheme 1c).

We initiated this novel sequential reaction with *N*-phenylpyrrolidine (**1a**) and dimethyl acetylenedicarboxylate (DMAD, **2a**), along with a variety of photocatalysts, inorganic bases, hydrogen acceptors, and light sources. To our delight, it was found that the desired [2+2] cycloaddition product **3aa** was produced in 70% yield when Ru(bpy)₃(PF₆)₂, nitrobenzene, and KOAc were used in the presence of a 23 W fluorescent light bulb upon photolysis for 96 h at 35 °C (Table 1, entry 1). To improve this result, we evaluated several nitrobenzene derivatives (Table 1, entries 2–4). It turned out that those with electron-withdrawing substituents on the benzene ring gave better results, and the highest reaction efficiency was observed with pentafluoronitrobenzene (PFNB), which furnished the desired dehydrogenation/[2+2] cycloaddition product in 99% yield. Control experiments demonstrated that both the light and the hydrogen acceptor were indispensable for this transformation (Table 1, entries 5 and 6), and both the photocatalyst and the inorganic base were necessary for an efficient reaction (entries 7 and 8). Finally, the relative configuration of product **3aa** was determined by the single-crystal X-ray diffraction, which showed that **3aa** had the *syn* configuration.^[15]

Having identified optimal conditions for this dehydrogenation/[2+2] cycloaddition reaction, we examined the scope of

Table 1: Optimization of the reaction conditions.^[a]



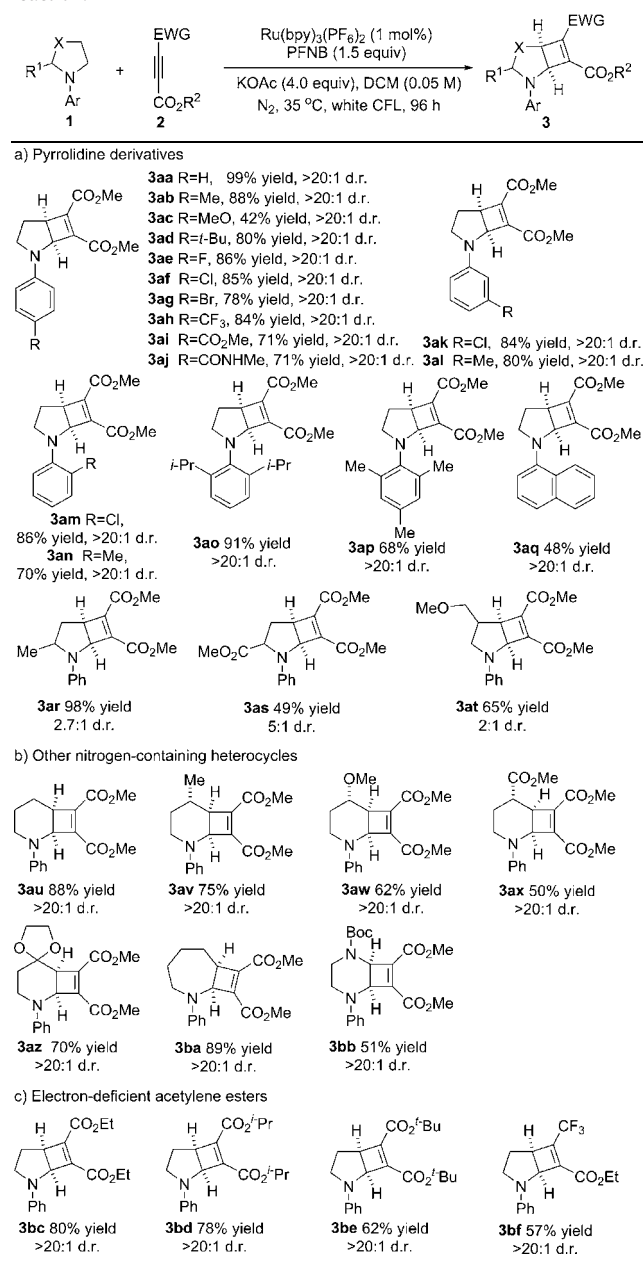
Entry	Hydrogen acceptor	Yield [%] ^[b]	d.r. ^[c]
1	nitrobenzene	70	> 20:1
2	1-methoxy-4-nitrobenzene	42	> 20:1
3	1-chloro-4-nitrobenzene	77	> 20:1
4	PFNB	99	> 20:1

Entry	Change from entry 4	Yield [%] ^[b]	d.r. ^[c]
5	no light	0	/
6	no PFNB	0	/
7	no Ru(bpy) ₃ (PF ₆) ₂	24	> 20:1
8	no KOAc	21	> 20:1

[a] Reactions were performed with **1a** (0.1 mmol), **2a** (0.3 mmol), the photocatalyst (1 mol%), a hydrogen acceptor (1.5 equiv), and KOAc (4.0 equiv) in DCM (2 mL) at 35 °C under a nitrogen atmosphere for 96 h. [b] Yield of the isolated product. [c] Determined by ¹H NMR spectroscopy. bpy = 2,2'-bipyridine, DCM = dichloromethane, PFNB = pentafluoronitrobenzene.

the reaction with respect to the *N*-aryl pyrrolidine cycloaddition partner. The sequential transformation gave the *syn* diastereoisomer as the major product in all cases studied owing to the high diastereoselectivity of the [2+2] cycloaddition process (Table 2). A range of *N*-aryl pyrrolidines bearing electron-donating or electron-withdrawing substituents at the *para* position of the benzene ring were suitable substrates (products **3ab–ag**). Some valuable functional groups most commonly found in drug molecules were also compatible with this system, including trifluoromethyl, ester, and amido groups (products **3ah–aj**). Both electron-rich and electron-deficient substituents at the *ortho* and *meta* positions of the phenyl ring were well tolerated (**3ak–ap**). *N*-Naphthylpyrrolidine was also successfully converted into the

Table 2: Scope of the sequential dehydrogenation/[2+2] cycloaddition reaction.



desired product **3aa** in good yield. The electronic nature of the substituents on the pyrrolidine ring played a critical role in the efficiency of this cycloaddition reaction: Product **3ar** bearing an electron-rich methyl group was obtained in 98% yield, whereas **3as** and **3at** with electron-withdrawing substituents were only produced in moderate yields (49 and 65%, respectively). Unfortunately, other pyrrolidine derivatives, such as *N*-methylpyrrolidine, *N*-(*tert*-butoxycarbonyl)pyrrolidine, and *N*-phenylpyrrolidin-2-one, did not furnish the corresponding product under the standard conditions.

The scope of the reaction with respect to the nitrogen-containing heterocyclic substrate was further evaluated. A range of *N*-phenylpiperidine derivatives smoothly delivered the corresponding [2+2] cycloaddition products in moderate to excellent yields, and only one diastereoisomer was obtained for compounds **3av–ax** owing to the high stability of this configuration (Table 2b). Furthermore, azepane and piperazine derivatives were found to be suitable substrates (products **3ba**, **3bb**). Finally, the alkyne component was examined in this cycloaddition protocol (Table 2c). A range of acetylene ester substrates could be smoothly transformed into the corresponding products **3bc–be** in good yields. Encouragingly, an unsymmetrically substituted acetylene ester substrate also furnished the [2+2] cycloaddition product **3bf** in 57% yield.

To demonstrate the applicability of this photocatalytic method in organic synthesis, we carried out the reaction on a 10 mmol scale under the irradiation of four 23 W fluorescent light bulbs. The cycloaddition product **3aa** was obtained in 97% yield (based on the recovered starting material) after two experiments (Scheme 2a). Furthermore, it was found that this versatile scaffold could be used to construct more complex polycyclic compounds through a Diels–Alder reaction (Scheme 2b). Under mild conditions, this scaffold was subjected to C(sp²)–H benzylation to install a valuable 2-hydroxy-5-nitrobenzyl moiety (Scheme 2c).

To further confirm our hypothesis, we conducted a series of mechanistic investigations (Figure 1). First, the quantum

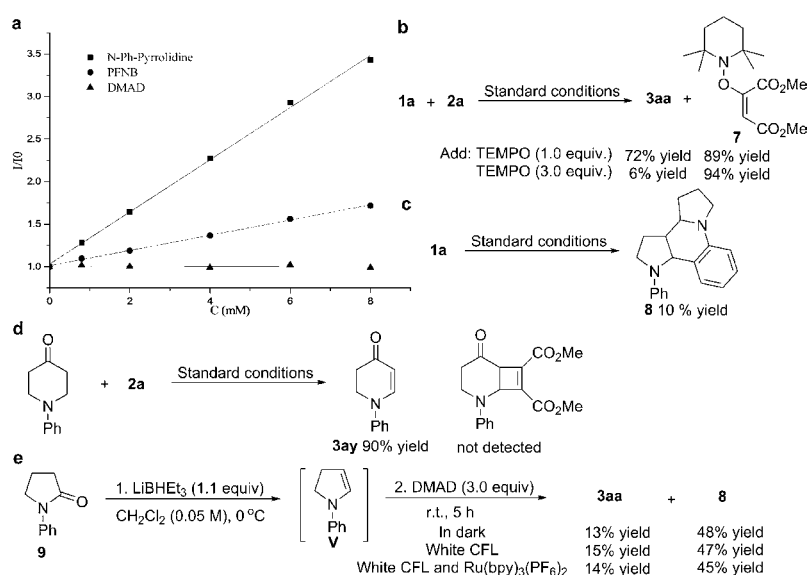
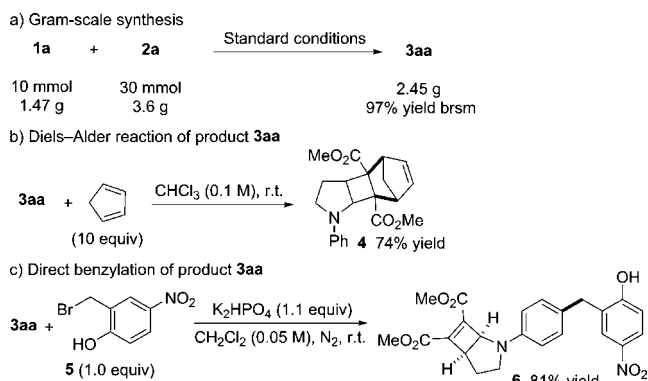


Figure 1. Mechanistic investigations.

yield of the reaction of **1a** and **2a** was determined to be 0.0002, which confirmed that the reaction was not a photo-initiated chain process, but a photocatalyzed process. Stern–Volmer experiments (Figure 1a) showed that the luminescence emission of excited-state $^*Ru(bpy)_3^{2+}$ was quenched by *N*-phenylpyrrolidine (**1a**) more efficiently than by pentafluoronitrobenzene (PFNB), which indicated a reduction quenching mechanism. Furthermore, under the standard conditions, when the radical quencher TEMPO was added and the amount was increased from 1.0 to 3.0 equivalents, the yield of product **3aa** sharply decreased, thus suggesting that this reaction involved a single-electron-transfer process (Figure 1b). Moreover, the dehydrogenation product **3ay** was obtained in 90% yield, which strongly demonstrated the high efficiency of this process (Figure 1d). Additionally, the fact that the dimerization product **8** was isolated in only 10% yield in the absence of DMAD (Figure 1c) indicated that both iminium ions and enamine intermediates **V** existed under these visible-light reaction conditions. Finally, a computational experiment for this [2+2] cycloaddition process provided three possible reaction pathways (see the Supporting Information). The fact that product **3aa** was obtained in similar yields in the dark and under irradiation with visible light (Figure 1e) supported the conclusion that the [2+2] cycloaddition between the enamine intermediate and DMAD was a thermal reaction (see the Supporting Information).

On the basis of a series of mechanistic experiments, we propose the mechanism shown in Figure 2 for this sequential visible-light photocatalyzed dehydrogenation/[2+2] cycloaddition reaction. It is well-known that $[Ru(bpy)_3]^{2+}$ (**I**) has a strong absorption cross-section in the visible range, and the excited state $^*[Ru(bpy)_3]^{2+}$ (**II**) ($^*Ru^{II}/Ru^{I} = 0.84$ V) will be highly populated through the acceptance of a photon from a variety of light sources. Subsequently, this high-energy intermediate **II** primarily undergoes single-electron transfer (SET) with the amine substrate **1a** ($E_{ox} = 0.74$ V versus SCE in CH₃CN) to initiate the first catalytic cycle and provide



Scheme 2. Application and transformation of product **3aa**.

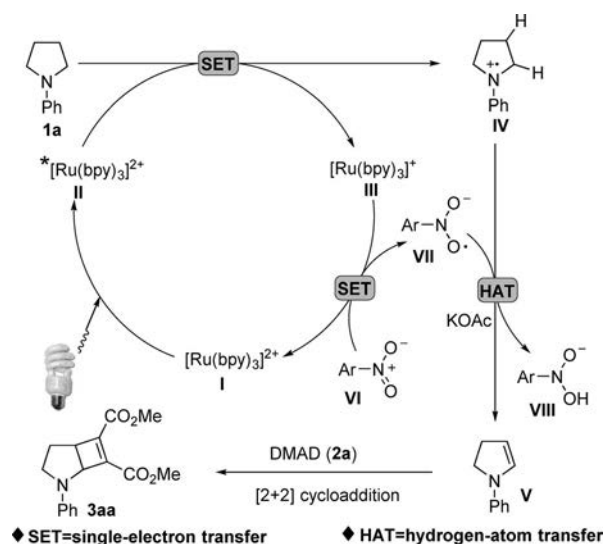


Figure 2. Proposed mechanism.

highly reducing $[\text{Ru}(\text{bpy})_3]^+$ (III) and the amine radical cation IV. Given that $[\text{Ru}(\text{bpy})_3]^+$ (III) has been shown to be a potent reductant ($\text{Ru}^{\text{II}}/\text{Ru}^{\text{I}} = -1.33$ V versus SCE in CH_3CN), a commercially available nitrobenzene VI (the reductive potential of PFNB is -0.96 V versus SCE in CH_3CN) serves as an electron and hydrogen acceptor to return $[\text{Ru}(\text{bpy})_3]^+$ (III) to the ground state and produce a highly activated nitrobenzene anion radical VII. By cooperation between the nitrobenzene anion radical VII and the weak base KOAc, the amine radical cation IV is effectively transformed into the desired enamine intermediate V. Finally, a thermal [2+2] cycloaddition reaction between the enamine intermediate V and dimethyl acetylenedicarboxylate (DMAD, 2a) successfully occurs to deliver the dual C–H functionalized product 3aa.

In summary, we have developed a novel visible-light photocatalyzed dehydrogenation/[2+2] cycloaddition sequence for the dual functionalization of two $\text{C}(\text{sp}^3)\text{--H}$ bonds. In this transformation, an array of elusive cyclic enamines were formed under mild conditions. Small strained cyclobutene skeletons attached to saturated N-heterocycles were successfully constructed in situ to furnish a variety of complex bicyclic amino acid derivatives through this sequential process.

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Conflict of interest

The authors declare no conflict of interest.

Keywords: C–H functionalization · [2+2] cycloaddition · dehydrogenation · synthetic methods · visible-light photocatalysis

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- [15] CCDC 1568396 (**3aa**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.

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